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A solution with a volume of 250.0 mL,

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contains 1.47 g NH_3 , and 3.95 g NH_4NO_3 .
a. What is the pH of this solution? b. If 12 mL of 0.12 M HCl solution is added to this solution, what will be the pH? $K_a(\text{NH}_4^+) = 1.8 \times 10^{-5}$

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+F

In chemistry, a mixture is a material made up of two or more different

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substances which are not chemically combined. A mixture is the physical combination of two or more substances in which the identities are retained and are mixed in the form of solutions, suspensions and colloids.. Mixtures are one product of mechanically blending or mixing chemical substances such as elements and compounds ...

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Mixture - Wikipedia

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In chemistry, an ionic compound is a chemical compound composed of ions held together by electrostatic forces

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termed ionic bonding. The compound is neutral overall, but consists of positively charged ions called cations and negatively charged ions called anions. These can be simple ions such as the sodium (Na^+) and chloride (Cl^-) in sodium chloride, or polyatomic species such as the ammonium ...

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Ionic compound - Wikipedia

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Logam alkali adalah golongan (kolom)

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dalam tabel periodik yang berisi unsur-unsur litium (Li), natrium (Na), kalium (K), rubidium (Rb), sesium (Cs), dan francium (Fr). Golongan ini terletak pada blok-s tabel periodik karena seluruh logam alkali memiliki elektron terluarnya pada posisi orbital-s: konfigurasi unsur/elektron ini tercermin pada sifat karakteristik mereka.

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